# Methyl 2-Methyldithiocarbazate in Heterocyclic Synthesis: Preparation of 2,5Disubstituted 1,3,4-Thiadiazoles, Bis(1,3,4-Thiadiazolium) Salts and Macrocycles containing 1,3,4-Thiadiazole Subunits. X-Ray Crystal Structure of 2,2'-Bis[4,5-dihydro-5-(2-hydroxyethylimino)-4-methyl-1,3,4-thiadiazole] 

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#### Abstract

Reaction of the iminophosphorane 2, derived from methyl 2-methyldithiocarbazate and triphenylphosphine, with carbon disulphide gives the mesoionic compound 3 , which reacts with $\alpha, \omega$-dihalogeno compounds to give bis-(1,3,4-thiadiazolines) 7-9. However, reaction with ethylenediamine or 1,4tetramethylenediamine leads to macrocycles 11 and 12. Methyl 2-methyldithiocarbazate, by sequential treatment with dicarboxylic acid chlorides and perchloric acid, leads to the bis-(1,3,4-thiadiazolium) salts 19-21, which undergo nucleophilic displacement of the methylthio group, upon reaction with $\beta$-substituted ethylamines, to give products 22-28. The crystal structure of compound 22 has been determined by X-ray crystallography. The crystal could be described in a $P 1$ cell, with parameters $a=11.2862(17), b=7.7409(8), c=4.2676(2) \AA, \alpha=93.441(5), \beta=91.703(6)$ and $\gamma=$ $108.780(12)^{\circ}$. The molecule presents an internal crystallographic inversion centre, and the terminal OH groups are disordered between two tetrahedral positions of the bonded carbon atom.


Macrocycles containing heterocyclic subunits have been shown to possess interesting chemical and biochemical properties. In view of the limited examples of 1,3,4-thiadiazole inclusion in a macrocyclic framework, ${ }^{1}$ we herein describe the synthesis and characterization of macrocycles containing 2,5-diamino-1,3,4thiadiazole subunits connected by ethylene and 1,4-tetramethylene bridges, as well as their open-chain counterparts.
On the other hand, heteroarene oligomers constitute a class of heterocyclic compounds of increasing interest owing to their biological properties. ${ }^{2}$ Methods for the preparation of heterocyclic dimers involve palladium-mediated oxidative dimerization of heteroarenes, ${ }^{3}$ and metal-catalysed cross-coupling of heteroaryl halides with Grignard reagents, ${ }^{4}$ boronic acids ${ }^{5}$ or stannylheteroarenes. ${ }^{6}$ Herein we report an efficient new method for the synthesis of the previously unreported bis-( $1,3,4$ - thiadiazole) derivatives, which is based on the sequential treatment of methyl 2-methyldithiocarbazate with dicarboxylic acid chlorides and perchloric acid.

## Results and Discussion

The iminophosphorane 2, readily available from methyl 2methyldithiocarbazate 1 and triphenylphosphine dibromide, ${ }^{7}$ reacts with carbon disulphide at room temperature to give the mesoionic 4 -methyl-5-methylthio-1,3,4-thiadiazolium-2-thiolate $\mathbf{3}$ as a crystalline solid in $81 \%$ yield. The IR spectrum of compound 3 shows a strong absorption band at $1362 \mathrm{~cm}^{-1}$ which can be attributed to exocyclic C-S stretching. ${ }^{8}$ The absence of isothiocyanate bands provides support for the formulation of compound 3 as a cyclic mesoionic structure rather than as a valence tautomer. The ${ }^{1} \mathrm{H}$ NMR spectrum displays two singlets, at $\delta 2.82$ and 3.79 , due to the $S$-methyl and $N$-methyl groups, respectively. In the ${ }^{13} \mathrm{C}$ NMR spectrum, the quaternary carbons appear at $\delta_{\mathrm{C}} 181.0(\mathrm{C}-2)$ and 167.8 (C-5), and the $S$-methyl and $N$-methyl group carbons appear at $\delta 19.6$ and 40.6 , respectively. The mass spectrum shows the expected molecular ion as the base peak and the fragmentation pattern is in accord with the proposed structure.

The conversion $\mathbf{2} \longrightarrow \mathbf{3}$ presumably involves an initial aza-Wittig-type reaction between the iminophosphorane 2 and carbon disulphide to give an isothiocyanate as a highly reactive intermediate, which undergoes cyclization to afford the mesoionic compound 3 .
Compound 3 undergoes regioselective $S$-alkylation and subsequent demethylation by the action of methyl iodide or benzyl bromide to give the corresponding 5 -alk ylthio-3-methyl-1,3,4-thiadiazoline-2-thiones 4 and 5. Compounds 3 and 4 undergo $S$-methylation upon reaction with Meerwein's reagent to yield the 3 -methyl-2,5-bis(methylthio)-1,3,4-thiadiazolium salt 6. Similarly, reaction of compound $\mathbf{3}$ with 0.5 mol equiv. of several $x, \omega$-dihalogeno compounds in chloroform produces acyclic dimers 7-9 in good yield, which undergo $S$-methylation upon reaction with Meerwein's reagent to give unstable salts. Only compound $\mathbf{1 0}$ could be isolated and stored without any sign of decomposition (Scheme 1).

Compound 3 also reacts with $\alpha, \omega$-diaminoalkanes, and the nature of the reaction product appears to depend on the length of the diamine. When the high-dilution reaction of compound 3 and ethylenediamine or tetramethylenediamine is carried out in chloroform at reflux temperature for 6 h , in the presence of triethylamine, the corresponding [ $5,5^{\prime}$ ]ylenediamino $\left[2,2^{\prime}\right]$ -ylenediiminobis-(2,3-dihydro-3-methyl-1,3,4-thiadiazolino)phanes 11 and 12 are obtained as crystalline solids in good yield. However, the use of hexamethylenediamine or octamethylenediamine, under the same reaction conditions, leads to dithiols 13 and 14 in good yield, according with the behaviour observed in the reaction of the related 3-methyl-2-methylthio-1,3,4-thiadiazolium cation with hydrazine, ethylenediamine and $o$-phenylenediamine. ${ }^{9}$ Reaction of compound $\mathbf{3}$ with tetramethylenediamine in methylene dichloride at reflux temperature for 3 h , in the presence of methyl iodide, gives the 5,5'-bis(methylthio) derivative 15, which by treatment with an ethanolic solution of tetramethylenediamine is converted into the macrocyclic compound 12 (Scheme 2), which clearly confirms the proposed structure $\mathbf{1 2}$ for the product of the reaction of compound $\mathbf{3}$ and tetramethylenediamine. Structure


Scheme 1 Reagents and conditions: i, $\mathrm{Ph}_{3} \mathrm{P}^{2} \cdot \mathrm{Br}_{2} ;$ ii, $\mathrm{CS}_{2}$, room temp., iii, RX ; iv, $\mathrm{Me}_{3} \mathrm{O}^{+} \mathrm{BF}_{4}^{-}$; $\mathrm{v}, \mathrm{BrRBr}$


Scheme 2 Reagents and conditions: $\mathrm{i}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{CHCl}_{3}$, reflux, 6 h ; ii, $\mathrm{Et}_{3} \mathrm{~N}$, MeI, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, reflux, 2 h ; iii, $\mathrm{H}_{2} \mathrm{~N}\left[\mathrm{CH}_{2}\right]_{4} \mathrm{NH}_{2}, \mathrm{EtOH}$, reflux

12 was easily assigned by NMR spectroscopy. The ${ }^{1} \mathrm{H}$ NMR spectrum in $\left[{ }^{2} \mathrm{H}_{6}\right]$ dimethyl sulphoxide displays two different triplets, at $\delta 2.82$ for the $\mathrm{CH}_{2} \mathrm{NH}$ and at $\delta 2.90$ for the $\mathrm{CH}_{2} \mathrm{~N}=$, while the central methylenes appear as a multiplet centred at $\delta$ 1.60 and the $N$-methyl appears as a singlet at $\delta 3.20$. The ${ }^{13} \mathrm{C}$ NMR spectrum shows the two expected peaks for quaternary carbons at $\delta_{\mathrm{C}} 159.2$ (C-5) and 162.0 (C-2), four peaks for the
methylenes at $\delta_{\mathrm{C}} 25.7$ and 27.9 (central $\left.\mathrm{CH}_{2}\right), \delta_{\mathrm{C}} 39.0\left(\mathrm{CH}_{2} \mathrm{NH}\right)$ and $\delta_{\mathrm{C}} 55.3\left(\mathrm{CH}_{2} \mathrm{~N}=\right)$, in addition to the N -methyl group carbon at $\delta_{\mathrm{C}} 34.4$. The proton and carbon chemical-shift values in structure 12 are consistent with those previously observed for the open-chain analogues. ${ }^{7.9}$

In the ${ }^{1} \mathrm{H}$ NMR spectrum of compound 15 , the $N$-methyl and $S$-methyl signals appear as singlets at $\delta 3.86$ and 2.67 , respectively, while the methylene groups appear at $\delta 1.79$ (central $\mathrm{CH}_{2}$ ) and $3.34\left(\mathrm{CH}_{2} \mathrm{~N}=\right.$ ). In the ${ }^{13} \mathrm{C} \mathrm{NMR}$ spectrum the quaternary carbons appear at $\delta_{\mathrm{C}} 152.0(\mathrm{C}-5)$ and 166.1 (C-2), and the methylene carbons occur at $\delta_{\mathrm{C}} 24.8$ (central $\mathrm{CH}_{2}$ ) and $50.7\left(\mathrm{CH}_{2} \mathrm{~N}=\right)$. The mass spectrum gave a parent peak at $m / z 376$, and the fragmentation pattern is in agreement with the proposed structure.

On the other hand, compound 1 undergoes acylation upon reaction with diacyl chlorides to give acyclic derivatives. Thus, reaction of compound 1 with oxalyl dichloride, terephthaloyl dichloride, and pyridine-2,6-dicarboxylic dichloride leads to compounds 16-18, respectively, in excellent yield ( $86-98 \%$ ). These compounds undergo cyclization with perchloric acidacetic anhydride in dry diethyl ether at room temperature to give the salts $\mathbf{1 9 - 2 1}$ in excellent yield ( $87-91 \%$ ). The ${ }^{1} \mathrm{H}$ NMR spectra of compounds $\mathbf{1 9 - 2 1}$ show two sharp singlets, at $\delta 3.15$ 3.20 and $4.23-4.26$, due to the $S$ - and $N$-methyl groups, respectively. In the ${ }^{13} \mathrm{C}$ NMR spectra, the quaternary carbons appear at $\delta_{\mathrm{C}} 163.4-165.6$ (C-2) and at $180.3-182.7$ (C-5), while the $S$ - and $N$-methyl groups carbons occur at $\delta_{\mathrm{C}} 20.6-21.1$ and 42.0-42.1, respectively. Electron-impact mass spectra show the expected $\left[\mathrm{M}^{+}-2 \mathrm{ClO}_{4}{ }^{-}\right]$ion peaks.
$2,2^{\prime}$-Bis-(1,3,4-thiadiazolium) salt 19 reacts with 2 -substituted ethylamines such as ethanolamine, cysteamine and 2-bromoethylamine, in the presence of equimolecular amounts of triethylamine, to give the functionalized $2,2^{\prime}$-bis-(5-ethylimino-4-methyl-1,3,4-thiadiazolines) 22-24 in excellent yield (86$91 \%$ ). Preparation of azide 25 was achieved from bromide 24 and sodium azide in dimethyl sulphoxide (DMSO), in $79 \%$ yield. Similarly, the salt 20 reacted with ethanolamine to give diol 26, and the salt 21 with propylamine and ethanolamine to give products 27 and 28 , respectively, in high yield (Scheme 3).

In order to identify unambiguously the structure and spatial disposition of the rings and side-chains of the reaction products, X-ray structure determination of crystalline compound 22 has been performed. Table 1 presents the main geometrical features (see Fig. 1). Bond distances in the ring agree with the values found from a search in the CSD ${ }^{10}$ for such fragments; seven hits were obtained from the four compounds ${ }^{11-14}$ with codes BIPTDZ, CIDSOX, GEXVAG and MPTZTD. The angular deformations within the ring follow the pattern found in the aforementioned literature: a planar ring, with the typical bond angle of $88^{\circ}$ at the sulphur atom, with almost a pentagonal value at $\mathrm{C}(2)$ and $\mathrm{N}(4)$, and angles of $c a .117^{\circ}$ at $\mathrm{N}(3)$ and $\mathrm{C}(5)$. The centrosymmetric molecular has a conformation similar to that of a fully extended chain (torsion angles either $c a .0^{\circ}$ or $c a$. $180^{\circ}$ ), except at the terminal OH group, which is disordered into two (partially occupied) positions. This disorder is peculiar in the sense that it follows a tetrahedral pattern, as the two positions occupied by the OH group correspond to those of the tetrahedron arrangement of $\mathrm{C}(8)$ so, although $\mathrm{H}(8 \mathrm{~A})$ is well defined in the difference synthesis, the other hydrogen atom does not appear, hidden by the disordered oxygen atom. In this way, the two oxygen positions form torsion angles of $c a . \pm 60^{\circ}$ with respect to the $\mathrm{N}(6)$ nitrogen (Table 1).

As far as the disorder is concerned, we think that the molecules assume a centrosymmetrical configuration with the OH group either at the $\mathrm{O}(9)$ or the $\mathrm{O}(10)$ position, in a $50: 50$ ratio. They pack strongly through the $\mathrm{O}(10)-\mathrm{H}(10)$ interaction, in extended chains, somehow along the $\langle 101\rangle$ direction. The


Scheme 3 Reagents and conditions: i, $\mathrm{ClOCCOCl}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}$; ii, terephthaloyl dichloride, toluene, reflux; iii, pyridine-2,6-dicarbonyl dichloride, toluene, reflux; iv, $\mathrm{Ac}_{2} \mathrm{O}-\mathrm{HClO}_{4}, \mathrm{Et}_{2} \mathrm{O}, 0^{\circ} \mathrm{C}$ to room temp.; $\mathrm{v}, \mathrm{H}_{2} \mathrm{~N}\left[\mathrm{CH}_{2}\right]_{2} \mathrm{X}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{EtOH}$, reflux

Table 1 Selected geometrical parameters ( $\AA,{ }^{\circ}$ )

| $\mathrm{S}(1)-\mathrm{C}(2)$ | $1.778(3)$ | $\mathrm{S}(1)-\mathrm{C}(5)$ | $1.749(4)$ |
| :--- | ---: | :--- | ---: |
| $\mathrm{C}(2)-\mathrm{N}(3)$ | $1.380(5)$ | $\mathrm{C}(2)-\mathrm{N}(6)$ | $1.270(4)$ |
| $\mathrm{N}(3)-\mathrm{N}(4)$ | $1.357(4)$ | $\mathrm{N}(3)-\mathrm{C}(11)$ | $1.447(5)$ |
| $\mathrm{N}(4)-\mathrm{C}(5)$ | $1.287(5)$ | $\mathrm{C}(5)-\mathrm{C}(5)^{i}$ | $1.437(4)$ |
| $\mathrm{N}(6)-\mathrm{C}(7)$ | $1.456(6)$ | $\mathrm{C}(7)-\mathrm{C}(8)$ | $1.494(6)$ |
| $\mathrm{C}(8)-\mathrm{O}(9)$ | $1.335(7)$ | $\mathrm{C}(8)-\mathrm{O}(10)$ | $1.217(8)$ |
| $\mathrm{C}(2)-\mathrm{S}(1)-\mathrm{C}(5)$ | $88.1(2)$ | $\mathrm{S}(1)-\mathrm{C}(2)-\mathrm{N}(6)$ |  |
| $\mathrm{S}(1)-\mathrm{C}(2)-\mathrm{N}(3)$ | $107.6(3)$ | $\mathrm{N}(3)-\mathrm{C}(2)-\mathrm{N}(6)$ | $128.5(3)$ |
| $\mathrm{C}(2)-\mathrm{N}(3)-\mathrm{C}(11)$ | $123.2(3)$ | $\mathrm{C}(2)-\mathrm{N}(3)-\mathrm{N}(4)$ | $123.8(3)$ |
| $\mathrm{N}(4)-\mathrm{N}(3)-\mathrm{C}(11)$ | $119.6(3)$ | $\mathrm{N}(3)-\mathrm{N}(4)-\mathrm{C}(5)$ | $117.2(3)$ |
| $\mathrm{S}(1)-\mathrm{C}(5)-\mathrm{N}(4)$ | $111.2(3)$ |  |  |
| $\mathrm{S}(1)-\mathrm{C}(5)-\mathrm{C}(5)^{i}$ | $115.9(3)$ | $\mathrm{N}(4)-\mathrm{C}(5)-\mathrm{C}(5)^{i}$ | $122.2(3)$ |
| $\mathrm{N}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | $121.9(3)$ | $\mathrm{C}(2)-\mathrm{N}(6)-\mathrm{C}(7)$ | $116.8(3)$ |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{O}(19)$ | $111.4(4)$ | $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{O}(9)$ | $115.9(5)$ |
|  | $118.0(4)$ |  |  |
| $\mathrm{S}(1)-\mathrm{C}(2)-\mathrm{N}(6)-\mathrm{C}(7)$ |  |  |  |
| $\mathrm{N}(6)-\mathrm{C}(2)-\mathrm{N}(3)-\mathrm{N}(4)$ | $-0.5(5)$ | $\mathrm{N}(3)-\mathrm{C}(2)-\mathrm{N}(6)-\mathrm{C}(7)$ | $-179.3(3)$ |
| $\mathrm{C}(2)-\mathrm{N}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | $-179.9(3)$ | $\mathrm{N}(3)-\mathrm{N}(4)-\mathrm{C}(5)-\mathrm{C}(5)^{i}$ | $-178.8(3)$ |
| $\mathrm{N}(6)-\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{O}(10)$ | $-58.5(7)$ | $\mathrm{N}(6)-\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{O}(9)$ | $58.0(6)$ |
|  |  |  |  |
| $\mathrm{O}(9) \cdots \mathrm{O}(10)^{i i}$ | $2.852(9)$ | $\mathrm{O}(9)-\mathrm{H}(9) \cdots \mathrm{O}(10)^{i i}$ | $156(-)$ |
| $\mathrm{O}(9) \cdots \mathrm{O}(10)^{i i i}$ | $2.830(8)$ | $\mathrm{O}(10)-\mathrm{H}(10) \cdots \mathrm{O}(9)^{i i i}$ | $175(12)$ |
| $\mathrm{H}(9) \cdots \mathrm{O}(10)^{i i}$ | $1.94(-)$ | $\mathrm{H}(10) \cdots \mathrm{O}(9)^{i i i}$ | $1.95(12)$ |
| $\mathrm{O}(9)-\mathrm{H}(9)$ | $0.97(-)$ | $\mathrm{O}(10)-\mathrm{H}(10)$ | $1.05(13)$ |

[^0]weaker interaction through $H(9)$ joins chains along c, alternating oxygen positions, and closing the interaction circuit around ( $0.5,1.0,0.5$ ). The internal symmetry of the molecule facilitates the positional disorder of the chains (or an inversion twinning), giving rise to the pattern shown in Fig. 2.

The identity of compounds 22-28 has also been ascertained by 2D-NMR, HCCOR and DEPT experiments. In the ${ }^{1} \mathrm{H}$ NMR spectra of compounds $\mathbf{2 2 - 2 8}$, the $N$-methyl group
appears as a singlet at $\delta 3.36-3.99$. For compound 22 the $N$ - and $X$-linked methylene groups appear as well resolved signals at $\delta$ 3.14 and 3.60 , respectively; for 23 they appear as two multiplets centred at $\delta 3.84$ and 3.14 , respectively; for dibromide 24 as a complex multiplet centred at $\delta 3.56$; and for diazide $\mathbf{2 5}$ as a triplet at $\delta 3.51$ and a broad singlet at $\delta 3.34$, respectively. Salient features of the ${ }^{13} \mathrm{C}$ NMR spectra are given in Table 2. The mass spectra showed the expected molecular ion peak and the

Table $2 \quad{ }^{13} \mathrm{C} \delta$-Values of 2,2'-Bis-(5-ethylimino-4,5-dihydro-4-methyl-1,3,4-thiadiazoles)

| Compound | NMe | $\mathrm{XCH}_{2}$ | $=\mathrm{NCH}_{2}$ | $\mathrm{C}(2)$ | $\mathrm{C}(5)$ | Others |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $\mathbf{2 2}^{a}$ | 35.5 | 61.1 | 60.1 | 135.9 | 154.3 |  |
| $\mathbf{2 3}^{b}$ | 35.9 | 46.7 | 51.7 | 142.2 | 166.9 |  |
| $\mathbf{2 4}^{c}$ | 35.9 | 32.2 | 59.1 | 137.0 | 156.7 |  |
| $\mathbf{2 5}^{c}$ | 35.7 | 51.8 | 57.0 | 136.9 | 156.7 | $128.2,131.0$ |
| $\mathbf{2 6}^{b}$ | 39.0 | 59.8 | 54.4 | 152.1 | 169.3 | $11.9,118.5,136.8,148.9$ |
| $\mathbf{2 7}^{d}$ | 35.6 | 24.1 | 59.8 | 145.2 | 157.1 | $124.2,140.9,146.5$ |
| $\mathbf{2 8}^{c}$ | 38.5 | 66.0 | 50.0 | 155.6 | 170.8 |  |

${ }^{a}$ In $\left[{ }^{2} \mathrm{H}_{6}\right]$ DMSO. ${ }^{b}$ In $\left[{ }^{2} \mathrm{H}_{6}\right]$ DMSO-TFA. ${ }^{c}$ In CDCl. ${ }^{d}$ In TFA-CDCl ${ }_{3}$.


Fig. 1 The molecular structure of the dimeric molecule 22, showing the atomic numbering used for the crystallographic analysis ${ }^{15}$


Fig. 2 Crystal packing of the title compound, showing the disordered OH group and the hydrogen interactions
fragmentation pattern was in good agreement with the proposed structures.

## Experimental

M.p.s were determined on a Kofler hot-stage apparatus and are uncorrected. IR spectra were recorded on a Nicolet-FT 5DX spectrometer. ${ }^{1} \mathrm{H}$ NMR data were obtained using a Bruker AC 200 instrument at 200 MHz , with $\mathrm{SiMe}_{4}$ as internal standard. ${ }^{13} \mathrm{C}$ NMR spectra were recorded on a Bruker AC 200 at 50

MHz. J-Values are in Hz . Electron-impact mass spectra were run on a Hewlett Packard -993 C spectrometer at an ionization potential of 70 eV . Elemental analyses were performed with a Perkin-Elmer 240 C instrument. TFA is trifluoroacetic acid.

4-Methyl-5-methylthio-1,3,4-thiadiazolium-2-thiolate 3.-A suspension of methyl 2-methyl-3-(triphenylphosphoranylidene)dithiocarbazate $2(19.83 \mathrm{~g}, 50 \mathrm{mmol})$ in carbon disulphide $\left(100 \mathrm{~cm}^{3}\right.$ ) was vigorously stirred at room temperature for 8 h . The solvent was removed under reduced pressure and the resulting solid was triturated with diethyl ether ( $50 \mathrm{~cm}^{3}$ ), filtered, washed with diethyl ether $\left(4 \times 20 \mathrm{~cm}^{3}\right)$ and recrystallized from chloroform to give the title compound in $81 \%$ yield, as yellow prisms, m.p. $141^{\circ} \mathrm{C}$ (Found: C, 27.0; H, 3.4; $\mathrm{N}, 15.7$. Calc. for $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{~S}_{3}$ : C, 26.95; H, 3.4; N, $15.7 \%$ ); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1} 1491,1464,1425,1362,1324,1253,1111,1004$, 990,927 and 648; $\delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 3.79(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN})$ and $2.82(3 \mathrm{H}, \mathrm{s}, \mathrm{MeS}) ; \delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO} ;\right.$ GATEDEC $) 181.0(\mathrm{~s}, \mathrm{C}-2)$, $167.8(\mathrm{~m}, \mathrm{C}-5), 40.6\left(\mathrm{q},{ }^{1} J 144.4, \mathrm{MeN}\right)$ and 19.6 (q, ${ }^{1} J$ 144.4, $\mathrm{MeS}) ; m / z(\%) 178\left(\mathrm{M}^{+}, 100\right), 164(7), 105(43), 102(22), 91$ (27), $88(5), 87(5), 76(31), 74(16), 73(33), 72(58), 59(26)$ and 47 (27).

General Procedure for the Preparation of 5-Alkylthio-3-methyl-1,3,4-thiadiazoline-2-thiones 4 and 5.-To a solution of 4-methyl-5-methylthio-1,3,4-thiadiazolium-2-thiolate 3 ( 1.40 g , 5 mmol ) in chloroform ( $35 \mathrm{~cm}^{3}$ ) was added the corresponding alkyl halide ( 5.25 mmol ). The solution was refluxed for 6 h and, on cooling, the solvent was removed under reduced pressure. The resulting material was triturated with diethyl ether ( 20 $\mathrm{cm}^{3}$ ), filtered off and recrystallized from ethanol. 3-Methyl-5-methylthio-1,3,4-thiadiazol-2(3H)-thione $4(93 \%)$ was obtained as yellow needles, m.p. $84^{\circ} \mathrm{C}$ (lit., ${ }^{1} 81-82^{\circ} \mathrm{C}$ ); $v_{\text {max }}(\mathrm{Nujol}) / \mathrm{cm}^{-1}$ $1462,1429,1360,1310,1258,1150,1123,1061,1007,974,909$ and $702 ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3} ;\right.$ GATEDEC $) 185.0\left(\mathrm{q},{ }^{3} J_{\mathrm{CNCH}} 5.3, \mathrm{C}-2\right)$, $156.2\left(\mathrm{q},{ }^{3} J_{\mathrm{CSCH}} 3.4, \mathrm{C}-5\right), 38.6\left(\mathrm{q},{ }^{1} J 142.7\right.$, MeN) and $15.4\left(\mathrm{q},{ }^{1} J\right.$ 143.1, MeS); $m / z(\%) 178\left(\mathrm{M}^{+}, 100\right), 164$ (10), 132 (5), 105 (23), 102 (11), 91 (18), $88(5), 87(4), 76(26), 73(18), 72(33)$ and 59 (17). 5-Benzylthio-3-methyl-1,3,4-thiadiazol-2(3H)-thione 5 $(89 \%)$ was obtained as yellow prisms, m.p. $77^{\circ} \mathrm{C}$ (Found: C, 47.3; H, 4.0; N, 11.1. Calc. for $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{~S}_{3}$ : C, 47.2; H, 4.0; N , $11.0 \%$ ); $v_{\max }($ Nujol $) / \mathrm{cm}^{-1} 1495,1472,1454,1350,1263,1126$, $1051,906,777,715$ and $701 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 7.35-7.33(5 \mathrm{H}, \mathrm{m}), 4.30$ ( $2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{~S}$ ) and $3.83(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right.$; GATEDEC) 185.4 (q, $\left.{ }^{3} J_{\mathrm{CNCH}} 3.4, \mathrm{C}-2\right), 154.6\left(\mathrm{t},{ }^{3} J_{\mathrm{CSCH}} 4.8, \mathrm{C}-5\right), 134.9(\mathrm{~m})$, 128.7 (dm), 128.1 (dtt, ${ }^{1} J 161.4,{ }^{3} J 6.4,{ }^{3} J 1.7$ ), 38.7 (q, ${ }^{1} J 142.8$, MeN ) and $37.7\left(\mathrm{tm},{ }^{1} J 144.0, \mathrm{CH}_{2} \mathrm{~S}\right) ; m / z(\%) 254\left(\mathrm{M}^{+}, 13\right), 163$ (2), 149 (3), 121 (4), 105 (2), 91 (100), 87 (6), 77 (7), 76 (14) and 73 (7).

3-Methyl-2,5-bis(methylthio)-1,3,4-thiadiazolium Tetrafluoroborate 6.-Method $A$. To a solution of 3-methyl-5-methylthio-1,3,4-thiadiazol-2 $(3 H)$-thione $4(0.53 \mathrm{~g}, 3 \mathrm{mmol})$ in dry methylene dichloride $\left(20 \mathrm{~cm}^{3}\right)$ was added trimethyloxonium tetrafluoroborate $(0.47 \mathrm{~g}, 3.15 \mathrm{mmol})$, and the mixture was stirred under reflux for 3 h . On cooling, the separated crystals were collected by filtration and recrystallized from dry chloroform ( $94 \%$ ).

Method B. To a solution of 4-methyl-5-methylthio-1,3,4-thiadiazolium-2-thiolate $3(0.53 \mathrm{~g}, 3 \mathrm{mmol})$ in dry methylene dichloride ( $20 \mathrm{~cm}^{3}$ ) was added trimethyloxonium tetrafluoroborate ( $0.47 \mathrm{~g}, 3.15 \mathrm{mmol}$ ). The reaction mixture was refluxed for 4 h and, on cooling, the crystals formed were separated and recrystallized from dry chloroform, to give the title compound as prisms in $92 \%$ yield, m.p. $76^{\circ} \mathrm{C}$ (Found: C, 21.4; H, 3.3; N, 10.0. Calc. for $\mathrm{C}_{5} \mathrm{H}_{9} \mathrm{BF}_{4} \mathrm{~N}_{2} \mathrm{~S}_{3}$ : C, 21.4; H, 3.2; N, $10.0 \%$ ); $v_{\text {max }}($ Nujol $) / \mathrm{cm}^{-1} 1499,1457,1430,1394,1322,1270,1126,1097$, $1061,992,966,935$ and $703 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}-\left[{ }^{2} \mathrm{H}\right] \mathrm{TFA}\right) 4.09(3 \mathrm{H}$, $\mathrm{s}, \mathrm{MeN}$ ), 2.99 ( $3 \mathrm{H}, \mathrm{s}, 2-\mathrm{SMe}$ ) and 2.78 ( $3 \mathrm{H}, \mathrm{s}, 5-\mathrm{SMe}$ ); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}-\mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{D}\right) 178.3(\mathrm{C}-2), 167.9(\mathrm{C}-5), 41.7(\mathrm{MeN})$, 20.3 (2-SMe) and 16.5 (5-SMe); $m / z(\%) 193\left(\mathrm{M}^{+}-\mathrm{BF}_{4}{ }^{-}, 20\right)$, 178 (100), 105 (18), 102 (11), 91 (10), 76 (11), 74 (6), 73 (13) and 72 (19).

General Procedure for the Preparation of 5,5'-Ylenedithiobis-(3-methyl-1,3,4-thiadiazoline-2-thiones) 7-9.-To a solution of 4-methyl-5-methylthio-1,3,4-thiadiazolium-2-thiolate 3 ( 1.40 g , 5 mmol ) in chloroform ( $35 \mathrm{~cm}^{3}$ ) was added the corresponding $\alpha, \omega$-dihalogeno derivative ( 2.5 mmol ). The solution was stirred at reflux temperature for 6 h and the solvent was evaporated off under reduced pressure. The resulting material was scratched with diethyl ether ( $20 \mathrm{~cm}^{3}$ ), filtered off, washed with diethyl ether ( $2 \times 15 \mathrm{~cm}^{3}$ ) and recrystallized from ethanol. 5,5'-Tetra-methylenedithiobis-[3-methyl-1,3,4-thiadiazol-2(3H)-thione] 7 ( $92 \%$ ) was obtained as prisms, m.p. $118-119{ }^{\circ} \mathrm{C}$ (Found: C, $31.5 ; \mathrm{H}, 3.6 ; \mathrm{N}, 14.7$. Calc. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{~S}_{6}: \mathrm{C}, 31.4 ; \mathrm{H}, 3.7$; $\mathrm{N}, 14.6 \%$ ); $v_{\text {max }}(\mathrm{Nujol}) / \mathrm{cm}^{-1} 1468,1412,1358,1261,1125,1049$, 1002, 906, 723 and $\left.710 ; \delta_{\mathrm{H}}\left({ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 3.78(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN})$, $3.21\left(2 \mathrm{H}, \mathrm{t}, J 6.15, \mathrm{CH}_{2} \mathrm{~S}\right)$ and $1.81\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right)$; $\delta_{\mathrm{C}}\left({ }^{2} \mathrm{H}_{6}\right]$ DMSO $) 184.4$ (C-2), 155.1 (C-5), $38.6(\mathrm{MeN}), 32.4$ $\left(\mathrm{CH}_{2} \mathrm{~S}\right)$ and $27.3\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right) ; m / z(\%) 382\left(\mathrm{M}^{+}, 2\right), 299(18)$, 297 (17), 276 (15), 219 (32), 204 (18), 178 (10), 177 (100), 163 (16), 149 (22), 147 (5), 143 (6), 129 (9), 120 (8), 105 (55), 104 (41), 102 (19), 91 (12), 90 (11), 88 (19), 87 (12), 83 (15), 76 (22), 73 (30) and 55 (44).
5,5'-[(Z)-But-2-enylenedithio]bis-[3-methyl-1,3,4-thiadiazole$2(3 \mathrm{H})$-thione $] 8(74 \%)$ was obtained as needles, m.p. $110-111^{\circ} \mathrm{C}$ (Found: C, 31.45; H, 3.2; N, 14.8. Calc. for $\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{~S}_{6}$ : C, 31.6; H, 3.2; N, $14.7 \%$ ); $v_{\text {max }}($ Nujol $) / \mathrm{cm}^{-1} 1464,1425,1352,1255,1118$, 1043, 1003, 900, 800, 725 and 704; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 5.80(1 \mathrm{H}$, dtd, $J 8.99,6.60, J 1.49,=\mathrm{CH}), 3.87\left(2 \mathrm{H}, \mathrm{dd}, J 6.43,1.49, \mathrm{CH}_{2}\right)$ and $3.85(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 185.4(\mathrm{C}-2), 153.8(\mathrm{C}-5), 127.7$ $(=\mathrm{CH}), 38.7(\mathrm{MeN})$ and $29.8\left(\mathrm{CH}_{2} \mathrm{~S}\right) ; m / z(\%) 380\left(\mathrm{M}^{+}, 19\right), 219$ (15), 217 (77), 206 (65), 201 (14), 164 (100), 163 (92), 146 (46), 145 (34), 141 (33), 131 (32), 118 (24), 105 (50), 91 (24), 87 (98), 86 (29), 85 (66), 76 (35), 73 (72), 72 (65), 59 (30), 58 (33) and 55 (26).

5,5'-(o-Xylidenedithio)bis-[3-methyl-1,3,4-thiadiazole-2(3H)thione] 9 ( $73 \%$ ) was obtained as prisms, m.p. $112-113{ }^{\circ} \mathrm{C}$ (Found: C, 39.0; H, 3.2; N, 12.9. Calc. for $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{~S}_{6}$ : $\mathrm{C}, 39.05$; H, 3.3; $\mathrm{N}, 13.0 \%$ ) ; $v_{\max }$ (Nujol $) / \mathrm{cm}^{-1} 1466,1423,1352,1261,1122$, $1051,1008,902,868,846,783,767,707$ and $690 ; \delta_{\mathbf{H}}\left(\mathrm{CDCl}_{3}-\right.$ TFA) 7.43-7.30 ( $4 \mathrm{H}, \mathrm{m})$, $4.50\left(4 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{~S}\right)$ and $3.87(6 \mathrm{H}, \mathrm{s}$, $\mathrm{MeN}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right.$-TFA) 185.6 (C-2), 155.5 (C-5), 133.5, 131.5 , 129.4, $39.2(\mathrm{MeN})$ and $35.3\left(\mathrm{CH}_{2} \mathrm{~S}\right) ; m / z(\%) 430\left(\mathrm{M}^{+}, 7\right), 267$ (4), 219 (3), 191 (12), 164 (12), 162 (8), 136 (30), 135 (100), 134 (11), 131 (11), 105 (19), 104 (29), 91 (23), 77 (15), 76 (7), 73 (22) and 58 (15).

5,5'-Tetramethylenedithiobis-(3-methyl-2-methylthio-1,3,4thiadiazolium) Bistetrafluoroborate 10.-To a solution of 5,5'-tetramethylenedithiobis-[3-methyl-1,3,4-thiadiazole-2(3H)thione] $7(0.57 \mathrm{~g}, 1.5 \mathrm{mmol})$ in dry methylene dichloride $\left(20 \mathrm{~cm}^{3}\right)$ was added trimethyloxonium tetrafluoroborate $(0.47 \mathrm{~g}, 3.15$ mmol ), and the resulting mixture was refluxed for 6 h . On cooling, the crystals formed were collected by filtration and recrystallized from methylene dichloride to give the title
compound 10 in $69 \%$ yield as prisms, m.p. $158-159^{\circ} \mathrm{C}$ (Found: C, 24.7; $\mathrm{H}, 3.3$; $\mathrm{N}, 9.5$. Calc. for $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{~B}_{2} \mathrm{~F}_{8} \mathrm{~N}_{4} \mathrm{~S}_{6}$ : C, 24.6; H , 3.4; N, $9.6 \%$ ); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1} 1491,1462,1413,1280,1060$, $941,862,771$ and $706 ; \delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 4.02(6 \mathrm{H}, \mathrm{s}, \mathrm{MeN}), 3.37$ $\left(4 \mathrm{H}, \mathrm{t}, J 6.1, \mathrm{CH}_{2} \mathrm{~S}\right), 2.96(6 \mathrm{H}, \mathrm{s}, \mathrm{MeS})$ and $1.86(4 \mathrm{H}, \mathrm{t}, J 6.1$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~S}\right) ; m / z(\%) 412\left(\mathrm{M}^{+}-2 \mathrm{BF}_{4}{ }^{-}, 2\right), 382(5), 277(4)$, 218 (15), 203 (3), 193 (3), 178 (100), 164 (24), 114 (13), 105 (82), 102 (24), 91 (31), 76 (34), 73 (43), 72 (80), 59 (32) and 55 (31).

General Procedure for the Preparation of Compounds 11-14.To a solution of 4-methyl-5-methylthio-1,3,4-thiadiazolium-2thiolate $3(0.71 \mathrm{~g}, 4 \mathrm{mmol})$ in chloroform ( $40 \mathrm{~cm}^{3}$ ) were added the appropriate $x, \omega$-diamino derivative $(4 \mathrm{mmol})$ and triethylamine ( $0.4 \mathrm{~g}, 4 \mathrm{mmol}$ ). The mixture was stirred at reflux temperature for 6 h and the resulting solid was filtered off and recrystallized from the appropriate solvent to yield compounds 11 and 12 or 13 and 14 , depending on the chain length of the diamino derivative used (Method A). Compound 12 was also prepared by reaction of $2,2^{\prime}$-tetramethylenediiminobis-(2,3-dihydro-3-methyl-5-methylthio-1,3,4-thiadiazole) $15(0.75 \mathrm{~g}$, 2 mmol ) with tetramethylenediamine ( $0.22 \mathrm{~g}, 25 \mathrm{mmol}$ ) in ethanol ( $20 \mathrm{~cm}^{3}$ ) at reflux for 24 h , followed by the work-up previously described.
[5,5']Ethylenediamino[2,2']ethylenediiminobis-(2,3-dihydro3 -methyl-1,3,4-thiadiazolo)phane 11 ( $58 \%$ ) was obtained as prisms, m.p. $250{ }^{\circ} \mathrm{C}$ (from propan-1-ol) (Found: C, 38.5; H, 5.3; $\mathrm{N}, 35.7$. Calc. for $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{~N}_{8} \mathrm{~S}_{2}$ : C, 38.45; H,5.2; $\mathrm{N}, 35.9 \%$ ); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1} 1653,1568,1449,1400,1354,1302,1172,1128$, 1078, 1068, 1035, 1005, 929 and 692; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right.$-TFA) $8.63(1 \mathrm{H}$, $\mathrm{br} \mathrm{s}, \mathrm{NH}), 4.05-3.85(5 \mathrm{H}, \mathrm{m})$ and $3.66\left(2 \mathrm{H}, \mathrm{q}, J 5.14, \mathrm{CH}_{2} \mathrm{NH}\right)$; $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right.$-TFA) $169.8(\mathrm{C}-2)$, $161.4(\mathrm{C}-5), 47.9\left(\mathrm{CH}_{2} \mathrm{~N}\right), 39.1$ $\left(\mathrm{CH}_{2} \mathrm{NH}\right)$ and $37.1(\mathrm{MeN}) ; m / z(\%) 312\left(\mathrm{M}^{+}, 1\right), 286(6), 173$ (82), 156 (17), 140 (21), 114 (87), 102 (40), 99 (50), 98 (20), 88 (21), 87 (20), 86 (100), 85 (17), 84 (10), 73 (13), 72 (26), 70 (24) and 69 (90).
[5,5'] Tetrame thylenediamino[2, ${ }^{\prime}$ ' tetrame thylenediiminobis-(2,3-dihydro-3-methyl-1,3,4-thiadiazolo)phane 12 ( $78 \%$, Method A, and $75 \%$, Method B) was obtained as prisms, m.p. $166^{\circ} \mathrm{C}$ (from EtOH ) (Found: C, 45.8; H, 6.5; N, 30.5. Calc. for $\mathrm{C}_{14} \mathrm{H}_{24} \mathrm{~N}_{8} \mathrm{~S}_{2}: \mathrm{C}, 45.6 ; \mathrm{H}, 6.6 ; \mathrm{N}, 30.4 \%$ ); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1} 3369$, $1626,1568,1462,1404,1375,1338,1255,1143,1068,1018,1001$, 932 and $711 ; \delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 3.92(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH}), 3.20(3 \mathrm{H}, \mathrm{s}$, $\mathrm{MeN}), 2.90\left(2 \mathrm{H}, \mathrm{t}, J 5.78, \mathrm{CH}_{2} \mathrm{~N}\right), 2.82\left(2 \mathrm{H}, \mathrm{t}, J 6.79, \mathrm{CH}_{2} \mathrm{NH}\right)$ and 1.66-1.53 (4 H, m); $\delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$ DMSO) $162.0(\mathrm{C}-2),{ }^{2} 59.2$ (C-5), $55.3\left(\mathrm{CH}_{2} \mathrm{~N}\right), 39.0\left(\mathrm{CH}_{2} \mathrm{NH}\right), 34.4(\mathrm{MeN}), 27.9$ and 25.7; $m / z(\%) 241$ (2), 227 (16), 199 (52), 185 (100), 184 (3), 173 (48), 170 (12), 168 (42), 160 (14), 130 (44), 128 (9), 116 (12), 114 (19), 102 (31), 100 (7), 88 (23), 86 (20), 85 (14), 73 (21), 72 (55) and 70 (85).

5,5'-Hexamethylenedïminobis-(4,5-dihydro-4-methyl-1,3,4-thiadiazole-2-thiol) 13 ( $78 \%$ ) was obtained as prisms, m.p. 171$173{ }^{\circ} \mathrm{C}$ (from EtOH) (Found: C, 38.4; H, 5.4; N, 22.2. Calc. for $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{~S}_{4}: \mathrm{C}, 38.3 ; \mathrm{H}, 5.35 ; \mathrm{N}, 22.3 \%$ ); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1}$ $1609,1457,1432,1417,1289,1239,1191,1092,990,934$ and 638; $\delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$ DMSO $) 3.49(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN}), 3.14\left(2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CH}_{2} \mathrm{~N}\right)$, $1.56\left(2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right)$ and $1.32\left(2 \mathrm{H}, \mathrm{brs}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right)$; $\delta_{\mathrm{c}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 167.8(\mathrm{C}-2), 165.3(\mathrm{C}-5), 49.7\left(\mathrm{CH}_{2} \mathrm{~N}\right), 36.5$ ( MeN ), $28.2\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right)$ and $25.5\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right) ; m / z(\%)$ 376 ( $\mathrm{M}^{+}, 88$ ), 375 (6), 343 (27), 230 (17), 216 (17), 196 (44), 188 (17), 174 (12), 161 (16), 151 (14), 148 (100), 147 (75), 146 (25), 139 (11), 128 (12), 116 (10), 111 (12), 102 (13), 97 (13), 87 (11), 83 (16), 73 (12), 72 (16) and 69 (31).
5,5'-Octamethylenediiminobis-(4,5-dihydro-4-methyl-1,3,4-thiadiazole-2-thiol) $14(81 \%)$ was obtained as prisms, m.p. $164^{\circ} \mathrm{C}$ (from EtOH) (Found: C, 41.6; H, 6.05; N, 20.75. Calc. for $\mathrm{C}_{14} \mathrm{H}_{24} \mathrm{~N}_{6} \mathrm{~S}_{4}: 41.6 ; \mathrm{H}, 6.0 ; \mathrm{N}, 20.8 \%$ ); $v_{\text {max }}(\mathrm{Nujol}) / \mathrm{cm}^{-1} 1603$, $1431,1412,1230,1109,992,932$ and $722 ; \delta_{\mathrm{H}}\left(\left[^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right)$ $9.84(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH}), 3.61(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN}), 3.13(2 \mathrm{H}, \mathrm{t}, J 6.88$,
$\left.\mathrm{CH}_{2} \mathrm{~N}\right), 1.58\left(2 \mathrm{H}\right.$, br s, $\left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right)$ and $1.28(4 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right) ; \quad \delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) \quad 168.0 \quad(\mathrm{C}-2), \quad 165.5$ (C-5), $49.5\left(\mathrm{CH}_{2} \mathrm{~N}\right), 37.3(\mathrm{MeN}), 28.5\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right)$, $28.1\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right)$ and $25.9\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right) ; m / z(\%) 404\left(\mathrm{M}^{+}\right.$, 21), 403 (32), 371 (6), 180 (13), 178 (100), 164 (13), 148 (85), 147 (51), 139 (25), 129 (11), 111 (10), 102 (10), 97 (8), 87 (7), 83 (5), 73 (12) and 72 (17).

## 2,2'-Tetramethylenediiminobis-(2,3-dihydro-3-methyl-5-

 methylthio-1,3,4-thiadiazole) 15.-To a solution of 4-methyl-5-methylthio-1,3,4-thiadiazolium-2-thiolate $3(1.40 \mathrm{~g}, 5 \mathrm{mmol})$ in methylene dichloride ( $35 \mathrm{~cm}^{3}$ ) were added tetramethylenediamine ( $0.22 \mathrm{~g}, 2.5 \mathrm{mmol}$ ), triethylamine $(0.51 \mathrm{~g}, 5 \mathrm{mmol})$ and methyl iodide ( $2.13 \mathrm{~g}, 15 \mathrm{mmol}$ ). The reaction mixture was refluxed for 3 h , the solvent was removed under reduced pressure and the resulting material was triturated with cold ethanol ( $10 \mathrm{~cm}^{3}$ ). The solid formed was filtered off and recrystallized from ethanol to give the title compound in $53 \%$ yield, as yellow prisms, m.p. $198-199{ }^{\circ} \mathrm{C}$ (Found: C, 38.3; H, 5.4; $\mathrm{N}, 22.2$. Calc. for $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{~S}_{4}$ : C, $38.3 ; \mathrm{H}, 5.35 ; \mathrm{N}, 22.3 \%$; $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1} 1606,1494,1460,1419,1346,1319,1300,1250$, $1197,1072,972,945,916,758,707$ and $700 ; \delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right)$ $3.86(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN}), 3.34\left(2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CH}_{2} \mathrm{~N}\right), 2.67(3 \mathrm{H}, \mathrm{s}, \mathrm{MeS})$ and $1.79\left(2 \mathrm{H}\right.$, br s, $\left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right) ; \delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 166.1(\mathrm{C}-2)$, $152.0(\mathrm{C}-5), 50.7\left(\mathrm{CH}_{2} \mathrm{~N}\right), 38.5(\mathrm{MeN}), 24.8\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\right)$ and $16.1(\mathrm{MeS}) ; m /=(\%) 376\left(\mathrm{M}^{+}, 9\right), 343(28), 329(5), 270(15), 229$ (3), 216 (7), 214 (11), 202 (5), 200 (5), 188 (11), 174 (29), 168 (6), 161 (12), 156 (6), 147 (17), 142 (19), 128 (100), 127 (67), 116 (9), 102 (14), 91 (10), 82 (3), 74 (9), 73 (7), 72 (8) and 69 (22).Dimethyl 3,3'-Oxalylbis-(2-methyldithiocarbazate) 16.-To a solution of methyl 2-methyldithiocarbazate 1 ( $0.41 \mathrm{~g}, 3 \mathrm{mmol}$ ) in dry methylene dichloride ( $30 \mathrm{~cm}^{3}$ ) at $0^{\circ} \mathrm{C}$, was added oxalyl dichloride ( $0.42 \mathrm{~g}, 3.3 \mathrm{mmol}$ ) and the resulting suspension was kept at room temperature for 2 h . The solid formed was collected by filtration and crystallized from ethanol to give the title compound in $94 \%$ yield, as yellow needles, m.p. $255^{\circ} \mathrm{C}$ (Found: C, 29.5; H, 4.2; N, 17.1. Calc. for $\mathrm{C}_{8} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{O}_{2} \mathrm{~S}_{4}$ : C, 29.4; H, 4.3; N, 17.2\%); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1} 3268,1700,1481,1464$, $1377,1356,1265,1167,1118,1024,960,827$ and $721 ; \delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]-\right.$ DMSO) $12.08(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}), 3.57(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN})$ and $2.45(3 \mathrm{H}, \mathrm{s}$, $\mathrm{MeS}) ; \delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 201.2(\mathrm{C}=\mathrm{S}), 156.6(\mathrm{C}=\mathrm{O}), 42.9(\mathrm{MeN})$ and $19.1(\mathrm{MeS}) ; m / z(\%) 326\left(\mathrm{M}^{+}, 3\right), 278(2), 230(3), 163$ (98), $115(31), 91(100), 76(7), 74(20), 73(28), 47(16)$ and $46(8)$.

General Procedure for the Preparation of Dimethyl 3,3'-Arenediyldicarbonylbis-(2-methyldithiocarbazates) 17 and 18.To a vigorously stirred solution of methyl 2-methyldithiocarbazate $1(2.04 \mathrm{~g}, 15 \mathrm{mmol})$ in toluene $\left(30 \mathrm{~cm}^{3}\right)$ was added the corresponding diacyl chloride ( 7.5 mmol ), and the resulting suspension was stirred at reflux temperature for 4 h . After cooling, the solid formed was separated by filtration, washed successively with water ( $7 \mathrm{~cm}^{3}$ ) and diethyl ether ( $7 \mathrm{~cm}^{3}$ ) and crystallized from ethanol.

Dimethyl 3,3'-terephthaloylbis-(2-methyldithiocarbazate) 17 ( $98 \%$ ) was obtained as prisms, m.p. $341^{\circ} \mathrm{C}$ (Found: C, $41.8 ; \mathrm{H}$, 4.6; $\mathrm{N}, 13.9$. Calc. for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~N}_{4} \mathrm{O}_{2} \mathrm{~S}_{4}$ : C, 41.8 ; $\mathrm{H}, 4.5 ; \mathrm{N}, 13.9 \%$; $v_{\max }($ Nujol $) / \mathrm{cm}^{-1} 3269,1662,1525,1498,1464,1354,1278,1116$, 1047, 964, 860 and $727 ; \delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 11.82(2 \mathrm{H}, \mathrm{s}, \mathrm{NH})$, $8.06(4 \mathrm{H}, \mathrm{s}), 3.68(6 \mathrm{H}, \mathrm{s}, \mathrm{MeN})$ and $2.47(6 \mathrm{H}, \mathrm{s}, \mathrm{MeS})$; $\delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 201.5(\mathrm{C}=\mathrm{S}), 163.7(\mathrm{C}=\mathrm{O}), 134.7,127.9,43.4$ $(\mathrm{MeN})$ and $18.9(\mathrm{MeS}) ; m /=(\%) 402\left(\mathrm{M}^{+}, 2\right), 354(9), 308$ (12), 307 (35), 306 (94), 246 (13), 220 (13), 219 (100), 218 (10), 217 (52), 191 (8), 157 (40), 146 (12), 130 (14), 129 (32), 102 (18), 91 (31), 76 (9), 48 (25) and 47 (30).

Dimethyl 3,3'-(Pyridine-2,6-dicarbonyl)bis(2-methyldithiocarbazate) $18(86 \%)$ was obtained as flakes, m.p. $138-140^{\circ} \mathrm{C}$ (Found: C, 38.5; H, 4.4; N, 17.3. Calc. for $\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}_{4}$ : C,
38.7 ; $\mathrm{H}, 4.25 ; \mathrm{N}, 17.35 \%$ ); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1} 3412,3167,1682$, $1505,1456,1360,1315,1263,1231,1145,1106,1084,1036,1003$, 964, $901,872,843,738$ and $680 ; \delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 12.12(2 \mathrm{H}, \mathrm{s}$, NH ), 8.37 ( $3 \mathrm{H}, \mathrm{br} \mathrm{s}$ ), 3.75 ( $6 \mathrm{H}, \mathrm{s}, \mathrm{MeN}$ ) and 2.48 ( $6 \mathrm{H}, \mathrm{s}, \mathrm{MeS}$ ); $\delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 201.9(\mathrm{C}=\mathrm{S}), 160.8$ (pyridine $\mathrm{C}-2$ ), 146.7 ( $\mathrm{C}=\mathrm{O}$ ), 140.6 (pyridine $\mathrm{C}-4$ ), 126.1 (pyridine $\mathrm{C}-3$ ), 43.7 ( MeN ) and $19.1(\mathrm{MeS}) ; m /=(\%) 403\left(\mathrm{M}^{+}, 8\right), 356(8), 308(12), 264(2)$, 237 (4), 193 (7), 163 (2), 135 (2), 120 (3), 105 (10), 103 (4), 91 (100), 77 (13), 76 (6), 73 (7) and 47 (12).

General Procedure for the Preparation of 2,2'-Bis-(4-methyl-5-methylthio-1,3,4-thiadiazolium) Diperchlorates 19-21.-To a suspension of the appropriate dimethyl $3,3^{\prime}$-diacylbis-(2-methyldithiocarbazate) $\mathbf{1 6}-\mathbf{1 8}(2.5 \mathrm{mmol})$ indry diethyl ether $\left(15 \mathrm{~cm}^{3}\right)-$ acetic anhydride $\left(10 \mathrm{~cm}^{3}\right)$ at $0{ }^{\circ} \mathrm{C}$ was added $70 \%$ perchloric acid $\left(0.52 \mathrm{~cm}^{3}, 6 \mathrm{mmol}\right)$. Thereafter, the reaction mixture was kept at room temperature for 48 h while being vigorously stirred. The solid formed was filtered off, washed with diethyl ether ( $2 \times 10 \mathrm{~cm}^{3}$ ), dried and crystallized from a suitable solvent.

2,2'-Bis-(4-methyl-5-methylthio-1,3,4-thiadia-olium) diperchlorate $19(87 \%)$ was obtained as prisms, m.p. $288^{\circ} \mathrm{C}$ (from MeCN ) (Found: $\mathrm{C}, 19.5 ; \mathrm{H}, 2.5$; N, 11.3. Calc. for $\mathrm{C}_{8} \mathrm{H}_{12} \mathrm{Cl}_{2}-$ $\mathrm{N}_{4} \mathrm{O}_{8} \mathrm{~S}_{4}: \mathrm{C}, 19.6 ; \mathrm{H}, 2.5 ; \mathrm{N}, 11.4 \%$ ); $v_{\text {max }}(\mathrm{Nujol}) / \mathrm{cm}^{-1} 1489,1457$, $1426,1385,1283,1141,1094,1019,958,870,700$ and 625 ; $\delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 4.23(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN})$ and $3.20(3 \mathrm{H}, \mathrm{s}, \mathrm{MeS})$; $m /=(\%) 292\left(\mathrm{M}^{+}-2 \mathrm{ClO}_{4}{ }^{-}, 2\right), 262(5), 131$ (2), 110 (12), 105 (36), 91 (4), 76 (28), 74 (10), 73 (100), 58 (23), 52 (57), 47 (27) and 46 (40).

2,2'-(p-Phenylene)bis-(4-methyl-5-methylthio-1,3,4-thiadiazolium) diperchlorate $20(91 \%)$ was obtained as prisms, m.p. $320^{\circ} \mathrm{C}$ (from MeCN) (Found: C, 29.7; H, 2.8; N, 9.9. Calc. for $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{Cl}_{2} \mathrm{~N}_{4} \mathrm{O}_{8} \mathrm{~S}_{4}: \mathrm{C}, 29.6 ; \mathrm{H}, 2.8 ; \mathrm{N}, 9.9 \%$ ); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1}$ $1496,1466,1406,1304,1294,1093,922,848,721$ and 624 ; $\delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 8.26(4 \mathrm{H}, \mathrm{s}), 4.23(6 \mathrm{H}, \mathrm{s}, \mathrm{MeN})$ and $3.15(6 \mathrm{H}$, $\mathrm{s}, \mathrm{MeS}) ; \delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 180.3$ (C-5), 163.4 (C-2), 130.1, 128.8, $42.1(\mathrm{MeN})$ and $21.1(\mathrm{MeS}) ; m /=(\%) 368\left(\mathrm{M}^{+}-2 \mathrm{ClO}_{4}{ }^{-}, 1\right)$, 338 (29), 322 (23), 306 (13), 262 (4), 233 (64), 218 (37), 217 (100), 186 (5), 157 (71), 156 (33), 146 (36), 130 (30), 129 (82), 128 (55), 120 (11), 102 (20), 76 (24), 73 (75), 72 (84) and 48 (48).

2,2'-(Pyridine-2,6-diyl)bis-(4-methyl-5-methylthio-1,3,4thiadiazolium) diperchlorate $21(87 \%)$ was obtained as prisms, m.p. $308{ }^{\circ} \mathrm{C}$ [from EtOH-MeCN (1:2)] (Found: C, 27.3; H, 2.8; $\mathrm{N}, 12.2$. Calc. for $\mathrm{C}_{13} \mathrm{H}_{15} \mathrm{Cl}_{2} \mathrm{~N}_{5} \mathrm{O}_{8} \mathrm{~S}_{4}: \mathrm{C}, 27.5 ; \mathrm{H}, 2.7 ; \mathrm{N}, 12.3 \%$; $v_{\text {max }}(\mathrm{Nujol}) / \mathrm{cm}^{-1} 1466,1457,1436,1335,1283,1136,1096,1032$, $994,918,823,783,647$ and $624 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}-\mathrm{TFA}\right) 8.45-8.38(2 \mathrm{H}$, $\mathrm{m}), 8.26(1 \mathrm{H}, \mathrm{dd}, J 8.83,6.67), 4.26(6 \mathrm{H}, \mathrm{s}, \mathrm{MeN})$ and $3.16(6 \mathrm{H}, \mathrm{s}$, $\mathrm{MeS}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}-\mathrm{TFA}\right) 182.7(\mathrm{C}-5), 165.6(\mathrm{C}-2), 145.7$ (pyridine $\mathrm{C}-2$ ), 141.0 (pyridine C-4), 124.8 (pyridine $\mathrm{C}-3$ ), 42.0 ( MeN ) and 20.6 (MeS); $m /=(\%) 369\left(\mathrm{M}^{+}-2 \mathrm{ClO}_{4}{ }^{-}, 3\right), 339$ (12), 307 (25), 234 (27), 219 (38), 218 (57), 158 (83), 157 (34), 135 (6), 130 (95), 129 (37), 105 (37), 103 (100), 91 (8), 76 (59) and 73 (86).

General Procedure for the Preparation of 2,2'-Bis-(5-ethyl-imino-4,5-dihydro-4-methyl-1,3,4-thiadia-oles) 22-24 and 26-28.-To a suspension of the appropriate $2,2^{\prime}$-bis-(4-methyl-5-methylthio-1,3,4-thiadiazolium) diperchlorate 19-21 ( 5 mmol ) in dry ethanol $\left(40 \mathrm{~cm}^{3}\right)$ were added triethylamine $(1.01 \mathrm{~g}, 10$ mmol ) and the corresponding amino derivative ( 10 mmol ) and the reaction mixture was stirred at reflux temperature for 6 h . The resulting solid was filtered off, washed successively with water ( $2 \times 10 \mathrm{~cm}^{3}$ ) and diethyl ether $\left(10 \mathrm{~cm}^{3}\right)$ and crystallized from the appropriate solvent.

2,2'-Bis-[4,5-dihydro-(2-hydroxyethylimino)-4-methyl-1,3,4thiadiazole] $22(89 \%)$ was obtained as yellow prisms, m.p. 227$228^{\circ} \mathrm{C}$ (from EtOH) (Found: C, 38.0; H, 5.2; N, 26.45. Calc. for $\left.\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{~N}_{6} \mathrm{O}_{2} \mathrm{~S}_{2}: \mathrm{C}, 38.0 ; \mathrm{H}, 5.1 ; \mathrm{N}, 26.6 \%\right) ; v_{\text {max }}(\mathrm{Nujol}) / \mathrm{cm}^{-1}$ $3301,1630,1612,1504,1421,1325,1269,1089,999,879,848$
and $725 ; \delta_{\mathbf{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 4.67(1 \mathrm{H}, \mathrm{t}, J 5.62, \mathrm{OH}), 3.60(2 \mathrm{H}$, td, $\left.J 6.03,5.62, \mathrm{CH}_{2} \mathrm{OH}\right), 3.36(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN})$ and $3.14(2 \mathrm{H}, \mathrm{t}, J$ 6.03, $\left.\mathrm{CH}_{2} \mathrm{~N}\right) ; \delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}\right) 154.3(\mathrm{C}-5), 135.9(\mathrm{C}-2), 61.1$ $\left(\mathrm{CH}_{2} \mathrm{OH}\right), 60.1\left(\mathrm{CH}_{2} \mathrm{~N}\right)$ and $35.5(\mathrm{MeN}) ; m / z(\%) 316\left(\mathrm{M}^{+}\right.$, 17), 285 (100), 214 (45), 195 (60), 182 (78), 171 (66), 154 (34), 153 (87), 127 (91), 110 (12) and 99 (41).

2,2'-Bis-[4,5-dihydro-5-(2-mercaptoethylimino)-4-methyl-1,3,4-thiadiazole] $23(86 \%)$ was obtained as yellow prisms, m.p. 245-248 ${ }^{\circ} \mathrm{C}$ (from EtOH) (Found: C, 34.5; H, 4.7; N, 24.0. Calc. for $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{~N}_{6} \mathrm{~S}_{4}$ : C, 34.5; H, 4.6; N, 24.1\%); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1}$ $1634,1508,1463,1321,1273,1024,1005,986,856$ and 710 ; $\delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{DMSO}-\left[{ }^{2} \mathrm{H}_{6}\right] \mathrm{TFA}\right) 3.94(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN}), 3.88-3.80$ $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{~N}\right), 3.20-3.09\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{~S}\right)$ and $1.24(1 \mathrm{H}, \mathrm{t}, J$ $7.29, \mathrm{SH}) ; \delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$ DMSO- $\left[{ }^{2} \mathrm{H}_{6}\right]$ TFA $) 166.9$ (C-5), 142.2 $(\mathrm{C}-2), 51.7\left(\mathrm{CH}_{2} \mathrm{~N}\right), 46.7\left(\mathrm{CH}_{2} \mathrm{SH}\right)$ and $35.9(\mathrm{MeN}) ; m / z(\%) 348$ ( $\mathrm{M}^{+}, 3$ ), 315 (3), 301 (9), 288 (10), 241 (30), 228 (18), 200 (16), 174 (23), 169 (25), 143 (8), 142 (100), 119 (40), 116 (44), 115 (28), 114 (25), 110 (28), 88 (31), 87 (27), 86 (26), 73 (42), 72 (50), 61 (32), 60 (63), 59 (65), 55 (44) and 47 (34).

2,2'-Bis-[5-(2-bromoethylimino)-4,5-dihy'dro-4-methyl-1,3,4thiadiazole] $24(91 \%)$ was obtained as yellow prisms, m.p. $172{ }^{\circ} \mathrm{C}$ (from EtOH) (Found: C, 27.1; H, 3.3; N, 18.9. Calc. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{Br}_{2} \mathrm{~N}_{6} \mathrm{~S}_{2}$ : C, 27.2; H, 3.2; N, $19.0 \%$ ); $v_{\max }(\mathrm{Nujol})_{6}^{\prime} \mathrm{cm}^{-1}$ $1614,1512,1432,1358,1329,1276,1214,1023,996,856,748$, 678 and $616 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 3.59(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN})$ and $3.64-3.48(4 \mathrm{H}$, $\mathrm{m}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3} ;\right.$ GATEDEC $) 156.7\left(\mathrm{tq},{ }^{3} J_{\mathrm{C}=\mathrm{NCH}} 7.42,{ }^{3} J_{\mathrm{CNCH}} 2.04\right.$ C-5), 137.0 (s, C-2), 59.1 (dddd, ${ }^{1} J 138.23,{ }^{1} J 135.33,{ }^{2} J 4.06,{ }^{2} J$ $\left.2.95, \mathrm{CH}_{2} \mathrm{~N}\right), 35.9\left(\mathrm{q},{ }^{1} J 141.36, \mathrm{MeN}\right)$ and $32.2\left(\mathrm{dddd},{ }^{1} J 153.74\right.$, $\left.{ }^{1} J 152.70,{ }^{2} J 4.86,{ }^{2} J 3.89, \mathrm{CH}_{2} \mathrm{Br}\right) ; m / z(\%) 444\left(\mathrm{M}^{+}+4,2\right)$, $442\left(\mathrm{M}^{+}+2,4\right), 440\left(\mathrm{M}^{+}, 2\right), 439$ (11), 362 (3) 360 (3), 349 (14), 348 (100), 347 (13), 346 (93), 276 (2), 274 (2), 182 (63), 153 (41), 127 (58), 113 (6), 110 (26), 109 (12), 108 (11), 99 (32), 82 (10) and 80 (12).

2,2'-(p-Phenylene)bis-[4,5-dihydro-5-(2-hydroxyethylimino)-4-methyl-1,3,4-thiadiazole] $26(84 \%)$ was obtained as yellow prisms, m.p. $247-249^{\circ} \mathrm{C}$ (from EtOH ) (Found: C, 49.0; H, 5.2; $\mathrm{N}, 21.4$. Calc. for $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{O}_{2} \mathrm{~S}_{2}$ : C, $49.0 ; \mathrm{H}, 5.1 ; \mathrm{N}, 21.4 \%$ ); $v_{\max }($ Nujol $) / \mathrm{cm}^{-1} 3285,1624,1462,1423,1362,1341,1291$, $1254,1090,1019,927,899,872,832$ and 729 ; $\delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$ DMSO- $\left[{ }^{2} \mathrm{H}_{6}\right]$ TFA $) 8.06(4 \mathrm{H}$, s), $3.99(3 \mathrm{H}$, s, $\mathrm{MeN}), 3.79(2 \mathrm{H}, \mathrm{t}, J 4.32)$ and $3.60(2 \mathrm{H}, \mathrm{t}, J 4.32)$; $\delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$ DMSO- $\left[{ }^{2} \mathrm{H}_{6}\right]$ TFA $) 169.2$ (C-5), $152.1(\mathrm{C}-2), 131.0$, 128.2, $59.8\left(\mathrm{CH}_{2} \mathrm{OH}\right), 54.4\left(\mathrm{CH}_{2} \mathrm{~N}\right)$ and $39.0(\mathrm{MeN}) ; m / z(\%)$ 392 ( $\mathrm{M}^{+}, 14$ ), 362 (20), 361 (100), 318 (34), 247 (10), 235 (10), 229 (56), 217 (32), 203 (20), 157 (32), 146 (30), 129 (32), 120 (14), 102 (15), 73 (12) and 72 (12).

2,2'-(Pyridine-2,6-diyl)bis-(4,5-dihydro-4-methyl-5-propy-limino-1,3,4-thiadiazole) $27(88 \%)$ was obtained as needles, m.p. $150{ }^{\circ} \mathrm{C}$ (from EtOH) (Found: C, 52.5; H, 6.15; N, 25.0. Calc. for $\mathrm{C}_{17} \mathrm{H}_{23} \mathrm{~N}_{7} \mathrm{~S}_{2}: \mathrm{C}, 52.4 ; \mathrm{H}, 5.95 ; \mathrm{N}, 25.2 \%$; $v_{\text {max }}(\mathrm{Nujol}) / \mathrm{cm}^{-1}$ $1645,1634,1587,1568,1532,1456,1373,1344,1315,1277,1157$, $1072,1032,928,808,775,746,731$ and $642 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 7.85$ ( $1 \mathrm{H}, \mathrm{d}, J 2.01$ ), $7.81(1 \mathrm{H}, \mathrm{s}), 7.72(1 \mathrm{H}, \mathrm{dd}, J 9.04,6.18), 3.64$ ( $6 \mathrm{H}, \mathrm{s}, \mathrm{MeN}$ ), $3.16\left(4 \mathrm{H}, \mathrm{t}, J 7.06, \mathrm{CH}_{2} \mathrm{~N}\right), 1.74(4 \mathrm{H}, \mathrm{qt}, J 7.34$, 7.06, $\left.\mathrm{CH}_{2} \mathrm{Me}\right), 1.02\left(6 \mathrm{H}, \mathrm{t}, J 7.34, \mathrm{CH}_{2} \mathrm{Me}\right) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 1.57 .1$ (C-5), 148.9 (pyridine $\mathrm{C}-2$ ), 145.2 (C-2), 136.8 (pyridine $\mathrm{C}^{\prime}-4$ ), 118.5 (pyridine C-3), $59.8\left(\mathrm{CH}_{2} \mathrm{~N}\right), 35.6(\mathrm{MeN}), 24.1\left(\mathrm{CH}_{2} \mathrm{Me}\right)$ and $11.9\left(\mathrm{CH}_{2} \mathrm{Me}\right) ; m / z(\%) 389\left(\mathrm{M}^{+}, 20\right), 374(2), 361(21), 360$ (100), 259 (6), 230 (29), 204 (40), 179 (3), 165 (51), 158 (4), 136 (5), 130 (30), 103 (27), 101 (2), 69 (31) and 55 (7).

2,2'-(Pyridine-2,6-diyl)bis-[4,5-dihydro-2-hydroxyethylimino)-4-methyl-1,3,4-thiadiazole] $28(91 \%)$ was obtained as needles, m.p. $237{ }^{\circ} \mathrm{C}$ (from EtOH) (Found: C, 45.6; H, 4.9; N, 25.1. Calc. for $\mathrm{C}_{15} \mathrm{H}_{19} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}_{2}: \mathrm{C}, 45.8 ; \mathrm{H}, 4.9 ; \mathrm{N}, 24.9 \%$ ); $v_{\max }(\mathrm{Nujol}) / \mathrm{cm}^{-1}$ $3256,1622,1603,1566,1539,1456,1365,1277,1096,1053,928$, $812,776,736$ and $643 ; \delta_{\mathrm{H}}\left(\mathrm{TFA}-\mathrm{CDCl}_{3}\right) 8.47-8.37(2 \mathrm{H}, \mathrm{m}), 8.28$ $(1 \mathrm{H}, \mathrm{dd}, J 8.97,6.52), 4.81(4 \mathrm{H}, \mathrm{s})$ and $4.13(10 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}(\mathrm{TFA}-$ $\left.\mathrm{CDCl}_{3}\right) 170.8(\mathrm{C}-5), 155.6(\mathrm{C}-2), 146.5$ (pyridine $\mathrm{C}-2$ ), 140.9
(pyridine $\mathrm{C}-4$ ), 124.2 (pyridine $\mathrm{C}-3$ ), $66.0\left(\mathrm{CH}_{2} \mathrm{OH}\right), 50.0$ $\left(\mathrm{CH}_{2} \mathrm{~N}\right)$ and $38.5(\mathrm{MeN}) ; m /=(\%) 393\left(\mathrm{M}^{+}, 7\right), 363$ (19), 362 (100), 277 (5), 262 (4), 259 (5), 248 (12), 230 (89), 205 (12), 204 (99), 186 (9), 165 (82), 162 (17), 157 (15), 149 (18), 147 (44), 135 (16), 132 (7), 130 (75), 121 (12), 115 (27), 103 (64) and 77 (16).

2,2'-Bis-[5-(2-azidoethylimino)-4,5-dihydro-4-methyl-1,3,4thiadia=ole] 25.-To a solution of 2,2'-bis-[5-(2-bromoethyl-imino)-4,5-dihydro-4-methyl-1,3,4-thiadiazole] 24 ( $0.88 \mathrm{~g}, 2$ mmol ) in dry DMSO $\left(30 \mathrm{~cm}^{3}\right.$ ) was added sodium azide ( 0.52 g , 8 mmol ), and the reaction mixture was stirred at $70^{\circ} \mathrm{C}$ for 12 h . On cooling, the solution was poured into ice-water $\left(100 \mathrm{~cm}^{3}\right)$, and the solid formed was collected by filtration, washed with water ( $3 \times 15 \mathrm{~cm}^{3}$ ), dried and crystallized from hexane to give the title compound in $79 \%$ yield, as yellow needles, m.p. $138^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 32.9 ; \mathrm{H}, 3.7 ; \mathrm{N}, 45.7$. Calc. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{~N}_{12} \mathrm{~S}_{2}$ : C, 32.8; H, 3.85; N, $45.9 \%$ ); $v_{\text {max }}(\mathrm{Nujol}) / \mathrm{cm}^{-1}$ $2106,1649,1462,1359,1319,1275,1240,1074,1033,1006$, 972, 858 and $718 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 3.60(3 \mathrm{H}, \mathrm{s}, \mathrm{MeN}), 3.51(2 \mathrm{H}, \mathrm{t}$, $\left.J 5.29, \mathrm{CH}_{2} \mathrm{~N}\right)$ and $3.34\left(2 \mathrm{H}\right.$, br s, $\left.\mathrm{CH}_{2} \mathrm{~N}_{3}\right) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 156.7$ $(\mathrm{C}-5), 136.9(\mathrm{C}-2), 57.0\left(\mathrm{CH}_{2} \mathrm{~N}\right), 51.8\left(\mathrm{CH}_{2} \mathrm{~N}_{3}\right)$ and 35.7 (MeN); $m / z(\%) 366\left(\mathrm{M}^{+}, 13\right), 338$ (4), 310 (65), 282 (3), 269 (6), 255 (2), 241 (5), 214 (4), 187 (7), 182 (18), 171 (30), 159 (3), $154(18), 153$ (100), 141 (6), 128 (12) 127 (54), 114 (2), 88 (5), 87 (3), 86 (4), 73 (16), 72 (10) and 69 (66).

Crystallographic Analysis of 2,2'-Bis-[4,5-dihydro-5-(2-hy-droxyethylimino)-4-methyl-1,3,4-thiadiazole] 22.-Crystal data. $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{O}_{2} \mathrm{~S}_{2}, M=316.40$, space group $P 1, a=11.2862(17)$, $b=7.7409(8), c=4.2676(2) \AA, \alpha=93.441(5), \beta=91.703(6)$, $\gamma=108.780(12)^{\circ}, D_{\mathrm{c}}=1.493 \mathrm{~g} \mathrm{~cm}^{-3}, Z=1$. Cell constants obtained from a least-squares fit using 32 reflexions up to $045^{\circ}$, $\mathrm{Cu}-\mathrm{K} x$ radiation.

Data collection and processing. A transparent yellow prism sample $(0.20 \times 0.13 \times 0.03 \mathrm{~mm})$ was used for the anslysis on a Philips PW1 100 diffractometer, with $\mathrm{Cu}-\mathrm{K} \times$ radiation, graphite monochromator $\omega / 20$ scans, bisecting geometry, $1 \times 1^{\circ}$ detector apertures, $1.6^{\circ}$ scan width and using 1 min per reflexion. Good stability for the sample checked every 90 min . of 1104 independent reflexions, up to $75^{\circ}$ in 0,954 were considered observed [ $3 \sigma(I)$ criterion].

Structure analysis and refinement. The structure was solved by Direct Methods ${ }^{16}$ in the $P 1$ space group and refined ${ }^{17}$ by leastsquares procedures for 128 parameters in the $P 1$ space group, as refinement in $P 1$ did not progress. The OH groups appear to be disordered, population parameter 0.50 ; the $\mathrm{H}(9)$ atom had to be kept fixed in the last cycles of refinement and the $H(8 B)$ atom, superimposed on the corresponding $O(9)$ and $O(10)$ positions, could not be located. Empirical absorption correction ( $\mu 34.92 \mathrm{~cm}^{-1}$ ) was applied, ${ }^{18}$ the maximum and minimum transmission factors being 1.370 and 0.597 . All the hydrogen atoms but the mentioned $\mathrm{H}(8 \mathrm{~B})$ were located by difference synthesis and they were isotropically included in the last cycles of refinement. An empirical weighting scheme, so as to give no trends in $\left\langle\mu \cdot \Delta^{2} F\right\rangle$ vs. $\langle | F_{\mathrm{o}}| \rangle$ and $\langle\sin \theta / i\rangle$ was introduced. The final shift/error was 0.02 , with maximum peak in the final $\Delta F$ of $0.32 \mathrm{e}^{-3}$. The maximum thermal factor was $U 11(010) 0.093(5) \AA^{2}$. The final $R$ - and $R_{w}$-value were 0.049 and 0.050 , respectively. All the calculations were performed on a VAX 6410 computer. The atomic scaattering factors were taken from the International Tables. ${ }^{19}$ Table 3 shows the final nonhydrogen atomic co-ordinates.*

[^1]Table 3 Final atomic co-ordinates

| Atom | $x$ | $y$ | $=$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{S}(1)$ | $0.0462(1)$ | $0.2679(1)$ | $0.2932(2)$ |
| $\mathrm{C}(2)$ | $0.2095(3)$ | $0.3717(5)$ | $0.3836(8)$ |
| $\mathrm{N}(3)$ | $0.2498(3)$ | $0.2519(4)$ | $0.5500(7)$ |
| $\mathrm{N}(4)$ | $0.1631(3)$ | $0.0897(4)$ | $0.6031(7)$ |
| $\mathrm{C}(5)$ | $0.0543(3)$ | $0.0785(4)$ | $0.4855(7)$ |
| $\mathrm{N}(6)$ | $0.2805(3)$ | $0.5278(4)$ | $0.3175(7)$ |
| $\mathrm{C}(7)$ | $0.2224(4)$ | $0.6381(5)$ | $0.1430(10)$ |
| $\mathrm{C}(8)$ | $0.3175(5)$ | $0.8107(6)$ | $0.0567(12)$ |
| *O(9) | $0.4116(4)$ | $0.7968(7)$ | $-0.1145(11)$ |
| $* \mathrm{O}(10)$ | $0.3791(7)$ | $0.9110(10)$ | $0.2750(16)$ |
| $\mathrm{C}(11)$ | $0.3780(4)$ | $0.2921(7)$ | $0.6690(12)$ |

Population parameters: $p p(\mathrm{O} 9)=p p(\mathrm{O} 10)=0.5$

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[^0]:    Italics stand for symmetry operation: $i=-x,-y, 1-z, i i=x, y, z-1$ and $i i i=1-x, 2-y,-z$

[^1]:    * Supplementary Data: (see Section 5.6.3. of Instructions for Authors, in the January issue). Bond lengths and bond angles, together with their standard deviations and anisotropic thermal parameters, have been deposited at the Cambridge Crystallographic Data Centre.

